

Granitic Rock as a Reliable Multinutrient Fertilizer

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Received January 04, 2021; Revised February 10, 2021; Accepted February 18, 2021

Abstract Granite is the main component of the continental crust that is exposed over ca. 15 % of the continents. It is composed mainly of aluminosilicate minerals, rich in a variety of elements/nutrients. In Egypt, it outcrops along the Red Sea margin, Southern Sinai, Aswan, and southern the Western Desert. Weathered, fractured, and defective granites are easy to mine and inexpensive. Additionally, the granitic slurry and rock dust which result from cutting and polishing during the granitic manufacturing processes are considered waste products causing a problem and additional industrial costs. Reusing this waste product would benefit the manufacturer. On the other hand, the world's popularity as well as the Egyptian continues to increase at an alarming rate. Consequently, the food demand increases substantially. Egypt has a plan, already in action, for desert reclamation, however, these soils are poor in many fundamental nutrients. Additionally, the natural fertilization of the traditionally planted lands in the Nile Valley and the Delta decreases gradually since the establishment of the High Dam. Moreover, the worldwide high inflation rates cause a huge increase in the chemical fertilizers prices. Therefore, there is an urge and demand for affordable, available, and nutrients rich fertilizers. Granite, is traditionally, used as a potassium fertilizer which could enhance soil quality. However, its nature as an insoluble silicate-based mineral has limited the plants' beneficiary from its fertilizer. During this study, we have designed a strategy by which choosing the most suitable granitic rock type, overcoming its solubility issues, and testing its effect on plants. The treated sample is a monzogranite from the Younger granite suite of the Arabian-Nubian Shield at Somr ElQaa area in the northern Eastern Desert. The monzogranite rocks are generally characterized with minimum silica content among the granitic rocks and roughly equal concentrations of potassium and calcium.

Keywords: granitic fertilizer, mineral fertilizer, soil enrichment, Multinutrient fertilizer

Cite This Article: Sherif Mansour, and Amr Elkelish, "Granitic Rock as a Reliable Multinutrient Fertilizer." *World Journal of Agricultural Research*, vol. 9, no. 1 (2021): 42-46. doi: 10.12691/wjar-9-1-7.

1. Introduction

Granite is the dominant rock type in continental crusts, which is exposed over ca. 15 % of the continents [1]. Unlike other mineral fertilizers which support soil with one or few elements (i.e., phosphate rocks, marble, or limestone), granites are composed mainly of aluminosilicate minerals, rich in a variety of elements/nutrients. The Egyptian granitic rocks, mainly exposed along the Red Sea coast, have Neoproterozoic age (875-535 Ma; [2]), affected by successive tectonic events extended from the Cambrian to the Miocene. This prolonged and active history caused these rocks to be intensively weathered and fractured which reduced their economic value for industrial purposes, yet, these granites are easy to mine, crush, mill, and inexpensive. An additional affordable source for these granites, to be used as fertilizers, is the granitic slurry and rock dust that result from cutting and polishing during the granitic manufacturing processes. The granitic slurry is considered a waste product causing a problem and additional

industrial costs because it has to be collected, wrapped, transported, and dumped in an environmentally friendly way. An alternative is to recycle this waste product and used it for mineral fertilizer production that could benefit the granite manufacturer, the environment, the agricultural wealth, and the national/international economy.

The studied sample is a monzogranite from the Younger granite suite of the Arabian-Nubian Shield (ANS) at Somr ElQaa area in the northern Eastern Desert (Figure 1).

Food demand increases significantly with increasing world popularity. Similarly, on the national level, the Egyptian population grow continuously at an alarming rate. Additionally, the natural fertilization of the traditionally planted soils in the Nile Valley and the Delta decreases gradually since the establishment of the High Dam. Therefore, Egypt has a running plan to reclaim new agricultural land from the desert, unfortunately, many of these soils are poor in several fundamental nutrients. Moreover, the worldwide high inflation rates cause a huge increase in the chemical fertilizers prices. Therefore, there is an increasing need and demand for affordable, abundant, and nutrients rich fertilizers.

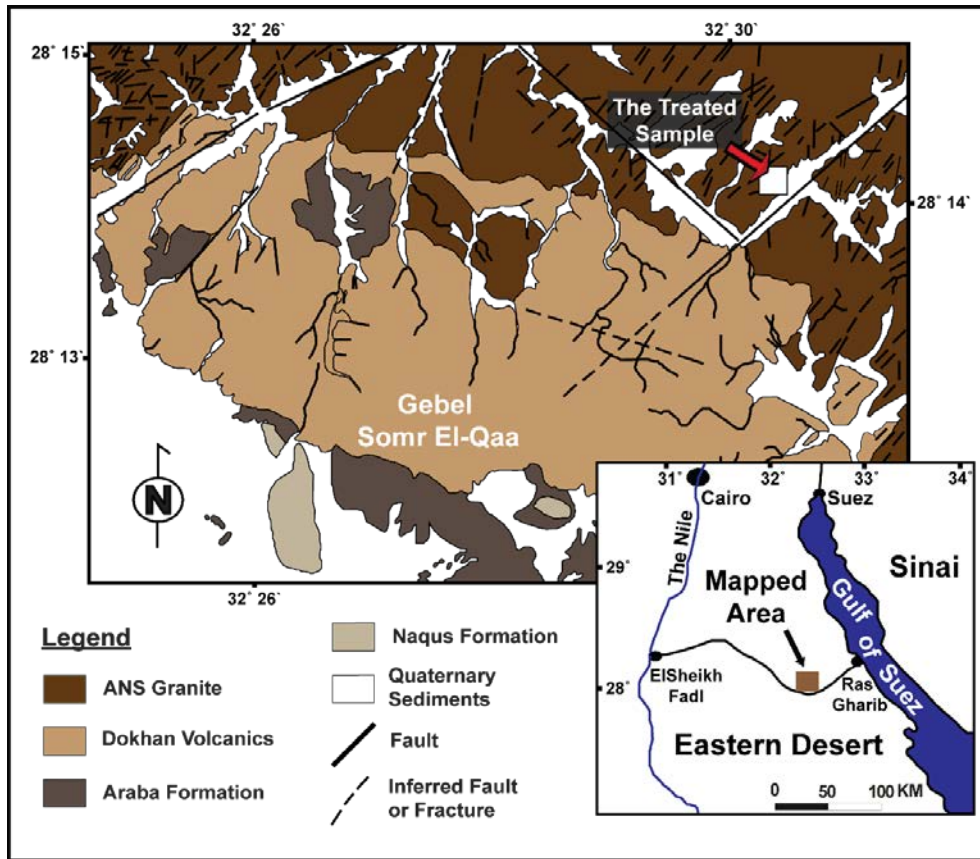


Figure 1. Geologic Map for Gebel Somr ElQaa area (modified after [4]), representing the main lithological units and the analyzed sample location

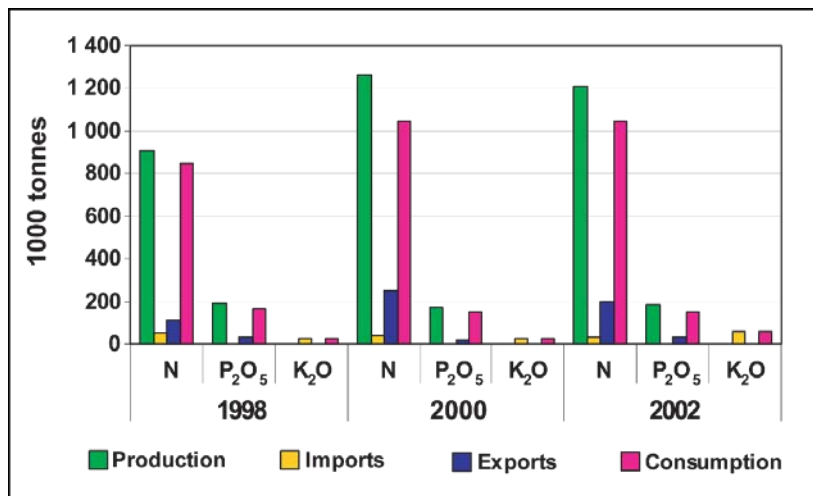


Figure 2. The production, imports, exports and consumption of fertilizers in Egypt [8]

In Egypt, mineral fertilizers along with organic manures are the main nutrient source for plants, especially the N, P and K fertilizers. These mineral fertilizers have traditionally been used in Egypt since 1902. The mineral fertilizers were entirely imported until national production of phosphate fertilizers began in 1936, while, nitrogen fertilizers production started in 1951. No potash fertilizers are produced in Egypt (Figure 2), due to the lack of resources [3]. The main types of fertilizers used in Egypt are: (1) as a Nitrogen nutrient; urea (46.5%), ammonium nitrate (33.5%), ammonium sulphate (20.6%), and calcium nitrate (15.5%). (2) as a Phosphate nutrient; single superphosphate (15%), and concentrated superphosphate (37%). (3) as a Potassium

nutrient; potassium sulphate (48-50%), and potassium chloride (50-60%) [3].

Granite, is traditionally, used as a potassium fertilizer which could enhance the soil quality (e.g., [5,6]). However, its nature as an insoluble silicate-based mineral has limited the plant's beneficiary from it as a fertilizer [7]. Therefore, finding the granitic rock type that is enriched in different nutrients, with low heavy metals and radioactive concentrations is the key, however, we still have to overcome its mineral solubility in media and/or soil issues and test its effectiveness on plants.

The analyzed sample is a monzogranite rock from the Younger granite suite of the ANS at Somr ElQaa area, northern Eastern Desert, Egypt (Figure 1). The

monzogranite rock sample was chosen because they are generally characterized with minimum silica content among granitic rocks, with roughly equal concentrations of potassium and calcium.

2. Treated Sample

The Egyptian ANS (ENS) was developed through multistage tectonic activities ended by island arcs and continental fragments accretion into old craton during the Late Neoproterozoic [9-13]. Granites with island-arc affinity are common in the ANS [14,15,16]. Previous studies proposed the classification of the ENS granitic rocks into two main groups; the older suite with Tonian-Cryogenian age and calc-alkaline composition, and the younger suite with Ediacaran age alkaline affinity [17,18,19]. However, more recent studies reported a synchronous formation of both calc-alkaline and alkaline granitic rocks [20,21]. After its formation, the ENS was eroded before the end of the Cambrian and buried beneath the Lower Palaeozoic sedimentary succession [22,23,24]. After a tectonically stable period, Gondwana and Laurasia collided causing the Hercynian tectonic events which caused rock uplift and exposure of the ENS again during the Devonian-Carboniferous [25,26,27]. Afterwards, the mid-Atlantic opening during the Jurassic-Cretaceous resulted in volcanism, domal structures, and additional exhumation in the ENS [28,29,30,31]. Then, during the Oligocene-Miocene, a period of significant uplifts in the ENS formed the flanks of the newly formed Red Sea/Gulf of Suez rift system.

The Younger granites formed in the ENS within two major events; (1) during the Dokhan event which is responsible for the emplacement of most of the ENS Younger granite in the Eastern Desert between 630 Ma and 570 Ma. (2) during the Katherina event which is responsible for the emplacement of most of the ENS Younger granite in the northern Eastern Desert and Sinai between 572 Ma and 500 Ma [32]. The analyzed sample was collected from the Younger granite that belongs to the

Dokhan event at Gebel Somr ElQaa area, western the Gulf of Suez, in the NED, Egypt (Figure 1). Generally, the Younger granites are characterized by SiO_2 concentrations ranges from 80 wt% and 70 wt%, the CaO ranges between 0.1 and 1.6 wt%, the $\text{K}_2\text{O}/\text{Na}_2\text{O}$ ratios > 1 , $\text{K}_2\text{O} > 3.8$ wt%, and the FeO^*/MgO ratios > 4 . They are enriched in total REEs and the high field strength elements, while, depleted in Ba, Sr, and have high Rb/Sr and conversely low K/Rb ratios [13].

3. Materials and Methods

The treated sample was collected from the Younger granite suite at Somr ElQaa area, and analyzed by conventional geological techniques; the boulder-sized sample are crushed using a Jaw crusher to reduce their size to ca. 200-75 mm. Then, a grinding Mill was used to reduce the size to the minimum achievable size. The resulting powder is insoluble because of the bonding of different elements with the silicate group (SiO_2). To overcome this issue, the resulting powder was treated with a mixture of concentric acids, these acids break the bond of different elements and dissolve them from the silicate component which did not dissolve. Then, the solution was filtered five times until it becomes entirely clear, while, the silicate group will be removed by this filtration process (Figure 3). The expected now is that the bond between different elements and the silicate group is broken, and all element except the silicate group will pass through the filter paper. To test the validity of this method we have analyzed the treated sample using the attenuated total reflection coupled with Fourier-transform infrared (ATR-FTIR). This technique is used to procure the infrared spectrum of the treated sample, it collects synchronously the high-spectral-resolution data over a widespread range. The resulting spectrum allows us to test whether the represented elements bonded to each other or not. Furthermore, the concentrations of the elements within the granitic powder and the granitic solution were examined using Energy Dispersive X-Ray Spectrometry (EDX).

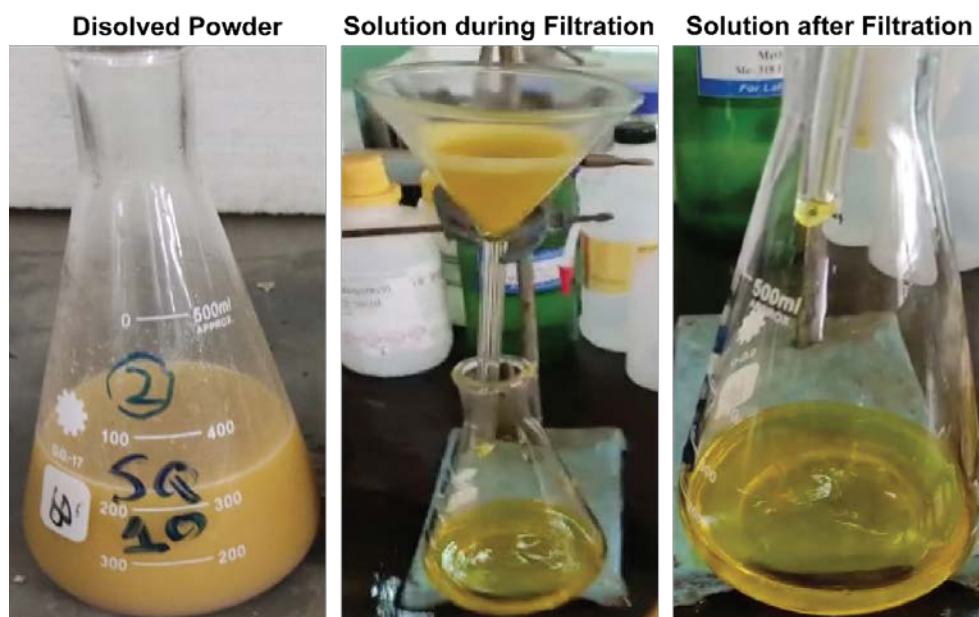


Figure 3. Dissolution and filtration processes of the treated sample

4. Results and Discussions

The studied sample was analyzed using EDX before and after being dissolved in acids to test if our method was efficient to remove the silicates from other elements. The powder (pre-solution) shows, as expected, a high silicate concentration of ca. 58.5% of the sample (Table 1). Then, descending concentrations of aluminium, iron, calcium, magnesium, sodium, and potassium oxides, indicating its suitability as a multinutrient fertilizer (Table 1). However, the main issue here is the insolubility of these elements and oxides in the water due to the insolubility of the silicate components which are connected to all other elements. Therefore, the element concentrations of the dissolved sample were measured using the same technique, EDX. The analyzed concentrations were very promising showing no silicate content after dissolving and filtering the undissolved silicates (Table 2). Taking into account that the detection limit of the EDX measurement is ca. 0.1%. This also is applied to the radioactive element which are absent or lower than our detection limit, documents for the absence of any radioactive hazardous for using these granites as mineral fertilizers.

Table 1. EDX analysis for the granitic powder

Element	wt%	σ	Atomic %	Oxide	Oxides %	σ
O	44.87	0.43	60.43			
Si	28.01	0.3	21.49	SiO ₂	58.47	0.43
Al	9.34	0.15	7.46	Al ₂ O ₃	17.33	0.18
Fe	6.34	0.33	2.45	FeO	8.29	0.30
Ca	3.74	0.69	2.01	CaO	5.14	0.64
Mg	2.91	0.08	2.58	MgO	4.75	0.09
Na	2.47	0.07	2.32	Na ₂ O	3.30	0.07
K	2.31	0.37	1.27	K ₂ O	2.72	0.29

Table 2. EDX analysis for the granitic solution

Element	wt%	σ	Atomic %
Cl	59.84	0.41	67.66
Fe	31.25	0.36	22.43
Al	3.34	0.16	4.96
Ca	1.64	0.15	1.64
K	1.34	0.15	1.37
Cr	1.33	0.18	1.02
Cu	1.10	0.30	0.69
P	0.17	0.12	0.22

Here we analyzed another granitic sample after dissolution because all granites are composed of silicate minerals, and the purpose of the dissolving process (removing silica) will be tested anyway. However, using another sample would demonstrate the fact that the granitoid rocks comprise a wide range of mineralogical and element compositions (Table 1 & Table 2). This is indicated here by representing not only different concentrations (from the powder sample) but also other elements. These elements represented in the dissolved granitic sample are chromium, copper, and phosphorus, while, the sodium element is absent. Therefore, further investigation is required to choose the most satisfactory rock that includes the most required nutrients.

5. Conclusion

Granitoid rocks can be used as multinutrient mineral fertilizer as it is rich in a majority of the required element for plants. However, further research about the most suitable rock type of event mixture of rocks is required. This result became achievable after we could overcome the main problem of using a granitoid rock as fertilizers by removing the insoluble silicate from the main component.

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